

ALKALOIDS OF BUXUS SEMPERVIRENS L. - VI (1) -
STRUCTURE OF BUXPSINE.

J. Tomko, O. Bauerova, Z. Voticky

(Institute of Chemistry, Slovak Academy of Sciences, Bratislava)

Alcaloïdes stéroïdiques - XLVIII (2)

(Famille des Buxacées ; 6^{ème} communication).

R. Goutarel and P. Longevialle

(Institut de Chimie des Substances Naturelles, Gif-sur-Yvette, France)

(Received 28 December 1965)

In a previous communication (1), the structure and absolute configuration of buxtauine and buxpiine, alkaloids extracted from Buxus sempervirens L., were reported. These are steroidal bases of a new type which have a single nitrogen atom, a cyclopropane ring, and a keto group in position 20.

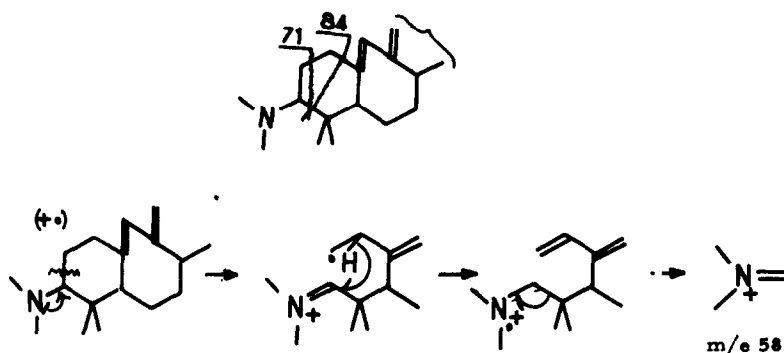
In the present note, we propose a structure for a new alkaloid to which we have given the trivial name buxpsiine (3).

Like buxpiine and buxtauine, buxpsiine I has a carbonyl group in position 20. However, it differs from these two bases by the presence of a conjugated double bond in position 16 and by the absence of a cyclopropane ring. The base has been isolated by chromatography on alumina of a mixture of alkaloids (fraction "C" (4)) originating from B. sempervirens.

Buxpsiine I has the formula $C_{26}H_{39}ON$ (calc. % C 81.83, H 10.36, N 3.67; found % C 81.96, H 10.79, N 3.52), molecular weight 381 (8), m.p. 176-178°, $[\alpha]_D^{24} + 105^\circ$ ($CHCl_3$, $c = 0.18$). The U.V. spectrum has maxima at 240 (log. ϵ 4.73) and 247m μ (log. ϵ 4.62), and a shoulder at 255m μ (log. ϵ 4.43). This spectrum is the sum of the absorptions of two chromophores: a conjugated ketone and a conjugated diene. The ORD curve shows a positive multiple Cotton effect, $[\alpha]_{350}^{25} + 1050^\circ$ corresponding to a conjugated carbonyl group. The I.R. spectrum has the two bands of an $\alpha\beta$ -unsaturated ketone at 1612 and 1690 cm^{-1} . Hydroxyl bands are absent. The NMR spectrum has singlets at δ 0.73ppm (6 protons; two tertiary methyls), δ 1.01ppm (3 protons; one tertiary methyl), δ 2.26ppm (9 protons; superimposed signals of the two groups $N(CH_3)_2$ and $COCH_3$), a signal with the appearance of a triplet centred on δ 6.64ppm (olefinic proton at position 16, β to a carbonyl group); a narrow signal ($\Delta\nu$ at half height 4cps) at δ 5.88ppm (olefinic proton at position 9a); and a multiplet centred on δ 5.53 ppm (olefinic proton at position 11).

These data show that buxpsiine I has three double bonds, one conjugated with the carbonyl group, the other two constituting a conjugated diene.

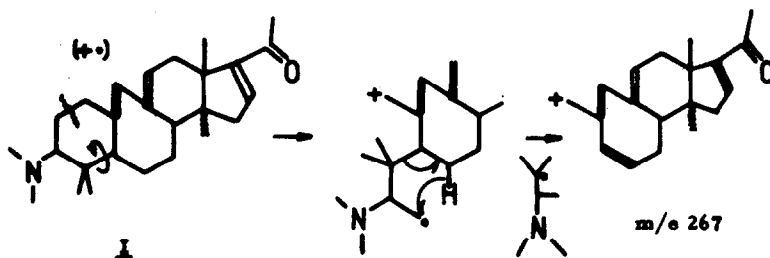
The mass spectrum of buxpsiine I reveals a molecular ion peak at m/e 381 (20% of the base peak). The dimethylamino group at position 3 is responsible for the presence of peaks at m/e 71 (base peak; 27% Σ_{27}) and m/e 84 (23% of the base peak), whose mechanisms of formation are well known (5), and a peak at m/e 58 (33%) for which the following mechanism of formation is proposed (6):



It may be noted that the conjugated diene renders more difficult the formation of the ion of mass 84 which is generally very important in the case of 3-dimethylamino steroids. On the other hand, the ion of mass 71 formed by homolytic rupture of the 1,2 bond is favoured. A similar situation is found in the series of Buxaceae alkaloids having a cyclopropane ring in position 9,10 (7).

Also noteworthy is the presence of a peak at m/e 43 (30% of the base peak) representing the ion $\text{CH}_3\text{C}\equiv\text{O}^+$, which comprises carbons 20 and 21, and a peak at m/e 338 (18% of the base peak) corresponding to the complementary ion (M-43) ($\text{M}^+ \rightarrow \text{M}-43$, metastable peak at m/e 299.8). The relative importance of this latter peak seems to be characteristic of Δ^{16} -20-oxo-pregnane derivatives (11).

The presence of a peak at m/e 336 (10% of the base peak ; M-45) is also of interest. A metastable peak at m/e 296.3 indicates the transition $M^+ \rightarrow M-45$. It is accompanied by peaks at m/e 321 (1%) (m/e 336 \rightarrow m/e 321 metastable peak at 306.6) and m/e 293 (1%) (m/e 336 \rightarrow m/e 293 metastable peak at 255.5) corresponding to the loss of the radicals, CH_3 and $\text{CH}_3\text{-}\dot{\text{C}}\text{O}$ (M-45-15 and M-45-43). The ion M-45 thus represents the loss of a molecule of dimethylamine from the molecular ion. Finally, there is a peak at m/e 267 (4%) which may be interpreted as follows :



The formation of this ion, stabilized by three conjugated double bonds, seems to be due to the presence of the diene and is a good confirmation of the proposed structure.

From these results, it is clear that buxpsiine **I** has the structure 3 ξ -dimethylamino-4, 4', 14 α -trimethyl-19 nor-B-homo-pregna-9a(10), 9(11), 16-trien-20-one. It is a new type of mono-amino steroidal alkaloid of the 19-nor-B-homo series, to which also belong the diamines buxamines or buxegines described by Stauffacher (9) and Kupchan (10).

References

1. Z. Voticky and J. Tomko, Tetrahedron Letters, 3579, (1965).
2. Alcaloïdes stéroïdiques - XLVII - Famille des Buxacées (5^{ème} communication), F. Khuong-Huu-Lainé, A. Milliet, N.G. Bisset et R. Goutarel, Bull. Soc. chim., in the press.
3. Z. Voticky : Communication to the International Symposium on the Chemistry and Stereochemistry of Indole and Steroid Alkaloids, Smolenice, 16 Septembre 1965.
4. J. Tomko, Z. Voticky, V. Paulik, A. Vassova, O. Bauerova, Chemicke Zvesti, 18, 721 (1964).
5. L. Dolejs, V. Hanus, V. Cerny, F. Sorm, Coll. czech. chem. Commun., 28, 1584 (1963).
6. W. Vetter, P. Longevialle, F. Khuong-Huu-Lainé, Q. Khuong-Huu et R. Goutarel, Bull. Soc. chim., 1324, (1963).
Z. Pelah, D.H. Williams, H. Budzikiewicz and C. Djerassi, J. amer. chem. Soc., 87, 574 (1965).
6. P. Longevialle, Thesis Dr. ès-Sci., Paris 1965.
7. D. Herlem-Gaulier, F. Khuong-Huu-Lainé, E. Stanislas, R. Goutarel, Bull. Soc. chim., 657, (1965).
8. The molecular weight was determined mass-spectrometrically by Dr. Dolejs, Czechoslovak Institute of Chemistry and Biochemistry, and Dr. Hanus, Czechoslovak Institute of Physical Chemistry.
9. D. Stauffacher, Helv. chim. Acta, 47, 968 (1964).
10. S.M. Kupchan, W.L. Asbun, Tetrahedron Letters, 3145 (1964)
11. The M-43 peak is more important in the spectra of pregn-16-en-20-one derivatives than in those of the saturated derivatives of pregna-20-one (cf. the examples in L. Peterson, Anal. Chem. 34, 1781 (1962)). This somewhat surprising fact has recently been emphasized by L. Tökes and C. Djerassi, Steroids, 6, 493 (1965).

